

Handling Volume Units in Chemistry

Volume unit questions – answers:

1) $0.013 \text{ dm}^3 = 13 \text{ cm}^3$

2) $2.42 \times 10^{-4} \text{ dm}^3 = 0.242 \text{ cm}^3$

3) $5.67 \times 10^{-2} \text{ cm}^3 = 5.67 \times 10^{-5} \text{ dm}^3$

4) $340 \text{ cm}^3 = 0.340 \text{ dm}^3$

5) $20 \text{ mL} = 20 \text{ cm}^3$

6) $4.6 \text{ mL} = 4.6 \times 10^{-3} \text{ dm}^3$

7) $86 \text{ }\mu\text{L} = 0.086 \text{ cm}^3$

8) $0.428 \text{ g cm}^{-3} = 428 \text{ g dm}^{-3}$

9) $7.45 \text{ g L}^{-1} = 7.45 \times 10^{-3} \text{ g cm}^{-3}$

10) $2.06 \times 10^{-5} \text{ mol dm}^{-3} = 2.06 \times 10^{-8} \text{ mol cm}^{-3}$